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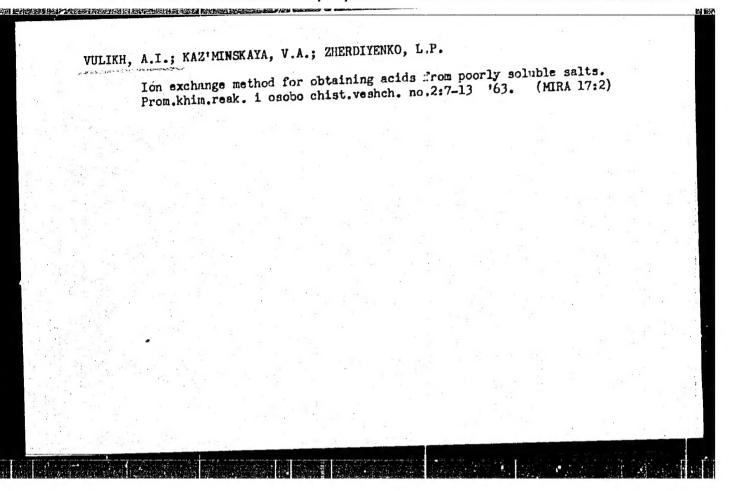
AUTHORS: Vulikh, A.I., Senyavin, M.M., Karoli, Yu.B., Korotkevich, B.I., Sidorova, L.G., and Galkina, N.K.

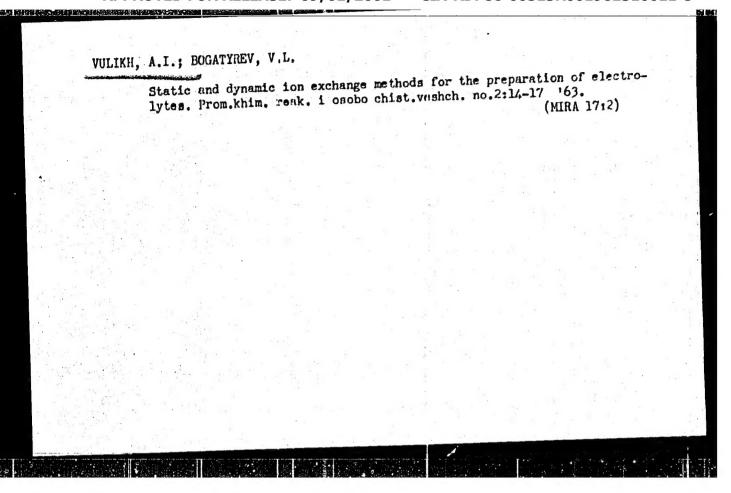
TITLE: A Method for Obtaining Rubidium or Cesium Carbonates

PERIODICAL: Byulleten' izobreteniy, 1960, Nr 10, p 19 Class 12 1, 16. Nr 128454 (622176/23 of Apr 21, 59).

TEXT: A method for obtaining rubidium or cesium carbonates from their dissolved salts, by a process of double ion-exchange. It has the following special features: to increase the degree of metal extraction, ammonium carbonate is used as the second reagent, whereupon the solution of rubidium (or cesium) carbonate and of ammonium carbonate is separated by distillation, while the solution or a salt separated therefrom is being heated.

Card 1/1





### "APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001961310011-8

VULIKH, A.I.; STATSENKO, A.A.; MAKOVETSKIY, M.I.; MIL'SKIY, S.A.

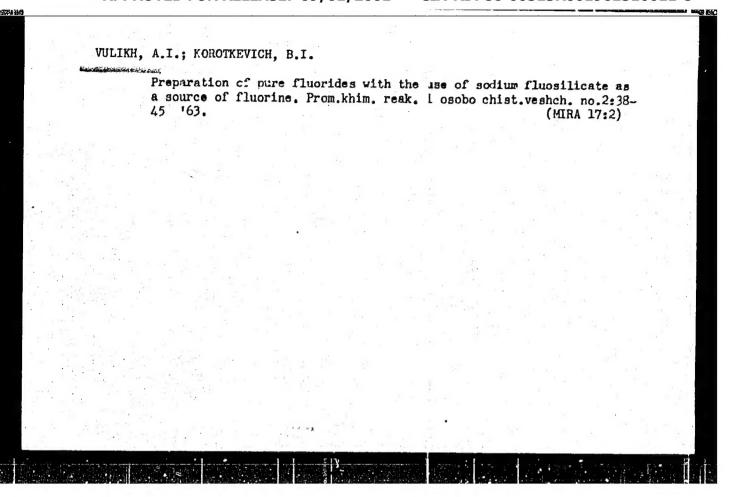
Chemical method for the preparation of welding fluxes. Prom.khim. reak.

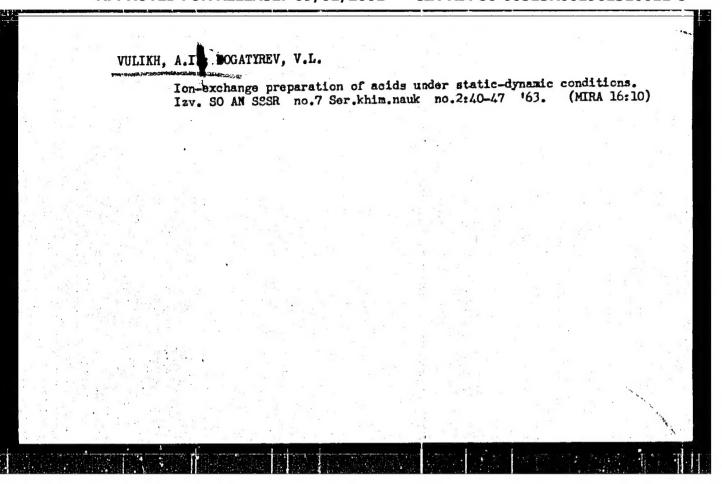
(MIRA 17:2)

1 osobo chist.veshch. no.2:18-22 '63.

L 12660-65 EWG( )/EWT(m)/RPM(c)/EPF(n)-2/EPR/EWP(b) Pr-4/Ps-4/Pa-ASD(d)/AS(mp)-2/AND(m)-3/AEDC(b) JD/JG/MLK ACCESSION AR: AT4C46116 8/0000/63/000/002/0023/0026 AUTHOR: Vulikh, A. I.; L.D. Prikhod ko; M. I. Hakiwatskiy TITLE: Preparation of aphydrous lithium hydroxide and oxide from lithium hydroxide monohydrate SOURCE: USSR. Gosudarstvenny\*y komitet khimicheskoy i neftyandy promy\*shlennosti. Promy\*shlennost' khimicheskiki reaktivov i osobo chisty\*kh veshchestv (Industry of chemical reagents and extra pure substances); informats founyty byulieten', no. 2. Moscow, IREA, 1963, 23 26 TOFIC TAGS: lithium hydroxide, lithium oxide, anhydrous lithium hydroxide; anhydrous lithium oxide, thermal decomposition, vacuum melting, vacuum dehydration, corundum crucible ABSTRACT: The thermal decomposition of lithium carbonate and lithium hydroxide monohydrate in a vacuum was investigated on a lorge scale, and the conditions for obtaining anhydrous lithium hydroxide and lithium oxide from the monohydrate were astablished. Among all the crucible materials tested, corundum was found to be the best for this purpose. A horizontal vacious electric furnace with a steel retort and Silit heaters was used, with a VN-2 oil racuum pump. The process was Card 1/2

at 300-350C in vacus (600- which is a porous product	first, the water of hyd at 650 mm Hg) and the anhydrius with a structure similar to age, the complete dissocilit	hydroxide was obtained, that of the initial cono-	
obtained at a gradually in decreasing pressure (down cake, which separates read lower than the theoretical side the crucible. Thus, lower than 900C, when the ide can be avoided. Chemi	creasing temperature (up to to 1 mm Hg). The resulting ily from the corundum cruit yield. No traces of the pr by remaining most of the wite vapor pressure of LiOH is at cal analysis showed that the	900-1000C) and a gradually	
			<u>.</u>
98-99% Li <sub>2</sub> 0, less than 0.1 hydroxide contained 0.5% C ASSOCIATION: None			
hydroxide contained 0.5% C		BUB CODE: IC, GC	

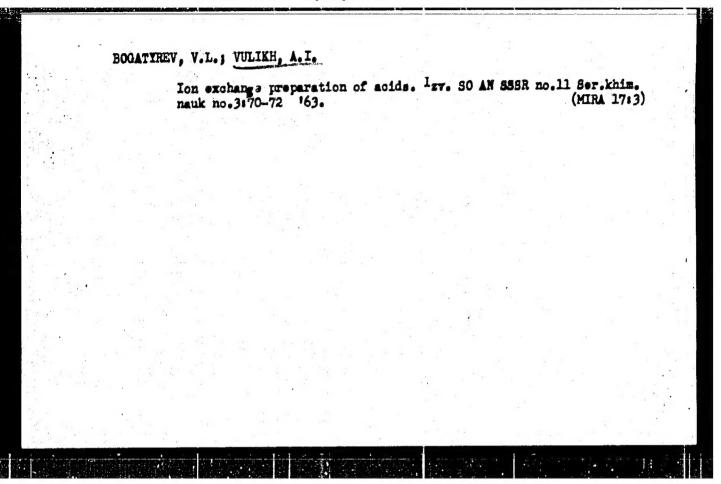




MIKULINSKIY, A.S.; KOZHEVNIKOV, G.N.; BAKHIREVA, L.D.; VULIKH, A.I.

Vacuum-thermal separation of cesium and potassium fluorides. Izv.
SO AN SSSR no.7 Ser.khim.nauk no.2:105-107 '63. (MIRA 16:10)

1. Ural'skiy filial AN SSSR, Sverdlovsk.



VULIKH, A.I., kand.tekhn.nauk; STATSENKO, A.A., inzh.; MAKOVETSKIY, M.I., inzh.; MIL'SKIY, S.A., inzh.

New technology for the production of fluxes for soldering and welding. Svar. proizv. no.9:24-26 S '63. (MIRA 16:10)

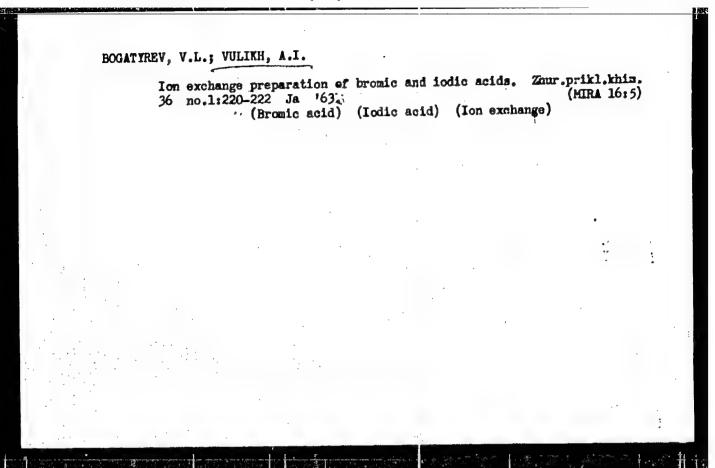
1. Novosibirskiy zavod khimicheskiki reaktivov.

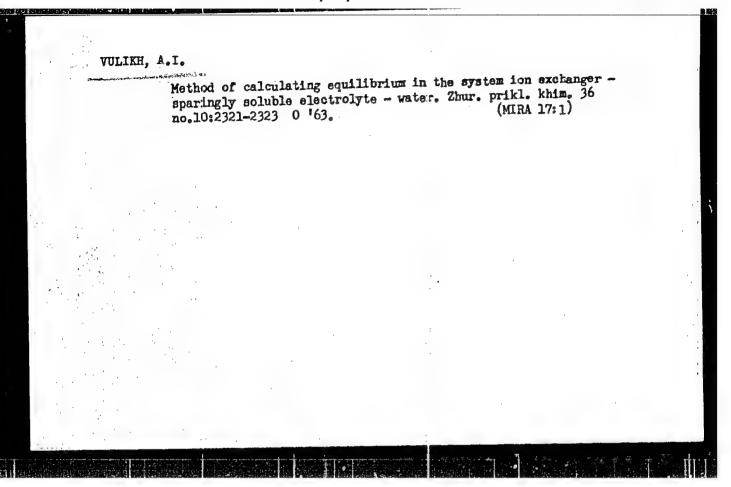
	N NRT /	PANNAN	5		s /	0286/63/0	00/020/00	12/0012
AUTHOR	Kuto11r	, s. A.	; Vul11	h. A.	I.			
No. 157	167		*		le le	titanates		
TOPIC T	AGS: ti n oxides	tanates, , titani	titan	ste syn kide, a	thesis, lkali no	alkali me etal hydro	etal titar oxide	of pre-
VDDIVVO	alkali m	etal tit	canaces	by fus	ing alk	ali metal	hydroxide	es with
paring titaniu	m dioxid			, 5,				
paring titaniu	m dioxid				CQ: 13D		ENCL:	00

CIA-RDP86-00513R001961310011-8" APPROVED FOR RELEASE: 09/01/2001

VULIKH, A.I. (Novosibirsk); KAZ'MINSKAYA, V.A. (Novosibirsk); ZHZRDIYENKO, L.P. (Novosibirsk)

Chemical experiments with the use of ion exchangers. Khim. v shkole 18 no.5:57-65 S-0 '63. (MIRA 17:1)





NIKOLAYEV, A.V.; BOGATYREV, V.L.; VULIKH, A.I.

Study of ion exchange processes by means of physicochemical analysis. Dokl. AN SSSR 153 no.2:360-362 N '63. (MIRA 16:12)

- 1. Institut neorganicheskoy khimii Sibirskogo otdeleniya AN SSSR. 2. Chlen-korrespondent AN SSSR (for Nikolayev).

CIA-RDP86-00513R001961310011-8" APPROVED FOR RELEASE: 09/01/2001

1/0076/64/008/010/2359/2361 ACCESSION NR: AP4046449 AUTHOR: Kutolin, S. A.; Druz', N. A.; Vulikh, A. I. TITLE: Second stable modification of lithium metazirconate SOURCE: Zhurnel neorganicheskiy khimit, v. 9 no. 10, 1964, 2359-2361 TOPIC TAGS: lithium metazirconate, stable midification, x ray analysis ABSTRACT: The existence of a second stable modification of lithium metazirconate in the Li20-ZrO2 system was established. Heating of 2LIOH ' ZrO2 in a muffie furnace for 4 hours at 950C gave a product LigZrO3 (II) which differed from that LizZr()3(1) synthesized by A. A. Grizik and V. Ye. Plyushchev (Zh. reorg. khimil, 6, 2249 (1961)) from Li<sub>2</sub>CO<sub>3</sub> + 2rO<sub>2</sub> by heating at 1100C for 1-2 Fours. Differences in the two modifications were established from x-ray data. No mutual transitions of the two modifications were noted. Differences in their hysical properties were established; fusion temperatures-Li2ZrO3I, 1900 ± 50C and Li, ZrOglf, 1530 ± 50C; Censities--4. 125 and 3. 508, respectively. The Card 1/2

ACCESSION NR: AP40464			
drolysis kinetics showed l LigZrO <sub>3</sub> I. Orig. art. has	1, ZrO3II hydrolyzed	much more readily than	
ASSOCIATION: None			
SUBMITTED: 02Jul63	ENCL: 0	0	
- SUB CODE: GC	NO REF SOV: 003	OTHER: 006	
Care2/2			
Constitution of the Consti	The state of the s		Annual process which will be seen as a first of the seen and the seen

NIKOLAYEV, A.V.; BOGATYREV, V.L.; VULIKH, A.I.

Ion-exchange system R<sup>+</sup>, NH<sup>+</sup>, R<sup>-</sup>, C.I.-H<sub>2</sub>O. Zhur. neorg. khim.

9 no.10:2469-2474, O '64.

(MIRA 17:12)

1. Institut neorganicheskoy khimii Sihirskogo otdeleniya AN SSSR.

L 231:85-65 EMT(#)/EMP(t)/EMP(b) INP(c) JD/30 ACCESSION NR: Al?5002192	QBQ/84/037/012/2748/2748
AUTHOR: Kutolin, S. A.; Vulikh, A. I.	
TITLE: Synthesis of alkali metal metatitanates un	ider vacium
SOURCE: Zhurnal prikladnov khimii, v. 37, no. 1	11, 1984, 2748
TOPIC TAGS: alkali metal metatitanate, synthes sium metatitanate	ii, lithium metatitanate, potas-
ABSTRACT: A method was worked out for the sy	ninesis of alkeli-metal titanates
was well mixed with a stoichiometric amount (Me.	uim. TiO2, "special grade",
brought to 650 or 800C for the Li or K. respective	retort. The temperature was
mospheric pressure. The pressure was then reduced temperatures were maintained for 2 more hours.	The friction tompositions of the
product $\mathrm{Li}_2\mathrm{TiO}_3$ was 1325 $\pm$ 50C and of $\mathrm{K}_2\mathrm{TiO}_3$ , 8 Card 1/2	20 ± 10C, Orig. art. has: no

"APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001961310011-8

L 2345-65			
ACCESSION NR: AP500219	2		
graphics			
ASSOCIATION: None			
SUBMITTED: 26Apr63	ENCL: 00	SUB CODE; (	C
NR REF SOV: 0//3	OTHER: 006		
Card2/2			

NIKOLAYEV, A.V.; BOGATYREV, V.L.: VULIKH, A.I.

Ion exchange system H°, Call, || R', C1 H<sub>2</sub>O investigated by the ray method. Dokl. AN SSSR 155 no. 3:607.-610 Mr 164.

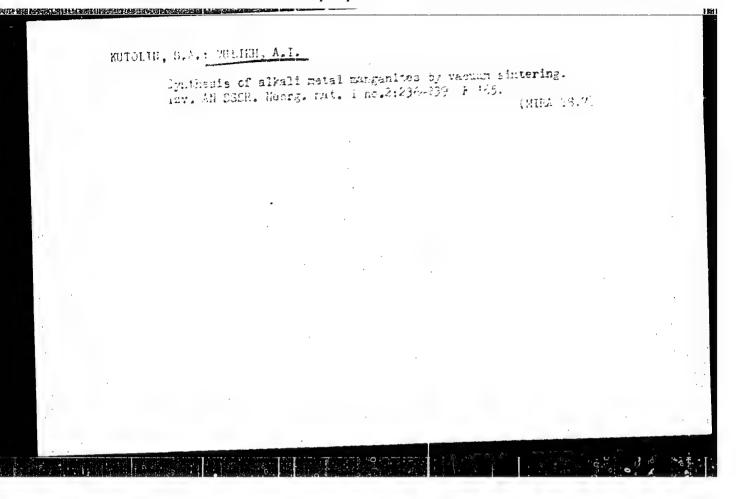
(MIRA 17:5)

1. Institut neorganicheskoy khimii Sibirskogo otdeleniya AN SSSR. 2. Chlen-korrespondent AN SSSR (for Nikolayev).

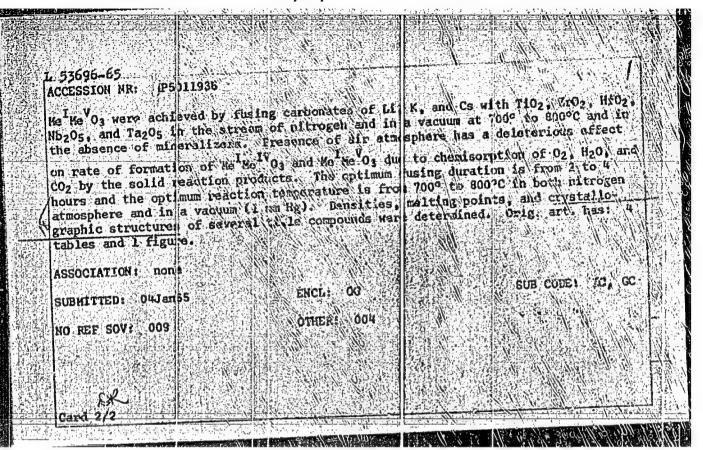
L 35064-65 ENT(m)/ENP(t)/EACCESSION HR: AP5008518			0019/0019 /5
AUTHOR: Vulikh, A. L.; Arkh:	promides and lodide	e of rubidium and ces	
Class 12, Ro. 169081  SOURCE: Byulleten' izobrete  TOPIC TAGS: reducing agent.	niv i tovarnykh zrakov	tio, 6, 1965, 19	nd, bromida,
iodide mui authoris Cer	tificate introduces a	jethod for preducing	bromides and
ABSTRACT: This Author's Cer iodides of rubidium and cesi carbonates or carbonates of reducing agents which produ- hydrazine hydrate as the re-	these metals with bro	mine or lodine in the	The use of
ASSOCIATION: none SUBHITTED: 11May64 HO REF SOV: 000	ENCL: 00 OTHER: OCO	SUS C	one. GC, IC
Card 1/1		The state of the s	

### "APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001961310011-8



L 53696-05 EMT(m)/EPF(n)-2/1/ESP(6)/ESP(b)/E ACCESSION NR: APSOLIGAGE  AUTHOR: Kutolin, S. A.; Vulikh, R. I.; Sergeyeva,  TITLE: Effect of atmospheres of various gases on  properties of Ma2 Ma 03 and Me 18 68 type compoun	the thermal synthesis and the D
TITLE: Effect of atmospheres of various gases on properties of Mar He Us and Me Iso type compound	the thermal synthesis and the
TITLE: Effect of atmospheres of various gases on aronauties of May Ma 03 and Me 18 03 type compound	the thermal synthesis and the
properties of Maz Ha O3 and Me 180 03 type	والإنتاج والمنافي والمنافية والمنافي
SOURCE: AN SSSR. Izvestiya. Neorganicheskiye mut	rialy, v. 1, no. 3, 1965, 358-391
TOPIC TAGS: thermal synthesis, mixed oxide salks	i metal, titanium, zicconium,
hafnium, niobium, tantalum, 1	s He 1 Ma VO 1 and Ma Me O3 (where
He is an alkali metal; He is Ti or , Hf in air and nitrogsn atmosphere; and in a vacuum. compounds is of interest to the nuclear power and The title compounds were propared by fusing mixtu	Preparation of these types of electrical caramic industries res of alkali metal carbonates with
oxides of the transition slepents in the tampera Fusing duration varies from 1 to 3 hours. Quanti Card 1/2	



L 34503-65 EW ACCESSION NR	T(m)/EMP(t)/EMP(b) IJP(c) JD, : AP5002802	0078/65/010/001/0140/0144
AUTHOR: Kuto	iln, S. A. Vulikh, A. L.	in vacuum
TITLE: Synthe	sis of alkali metal metatitanates	0, no. 1, 1965, 140-144
TOPIC TAGS:	alkali metal metatitanate, synth	esis, structure, density, fusion
temperature, i	iydrolysis	WO at atmospheric
pressure and t	der I min in the design of the state of the	12TiO3 and K2TiO3 were produced
quantitatively vacuum, High	er temperalures resulted in col corcelain crucibles. There was	no reaction between the carbonate
and TiO, in all hydroxide, X	r: under vacuum the reaction w rays showed the structure of th	s essentially the same as with the Li <sub>2</sub> TiO <sub>3</sub> was ordered. The densi
Card1/2		

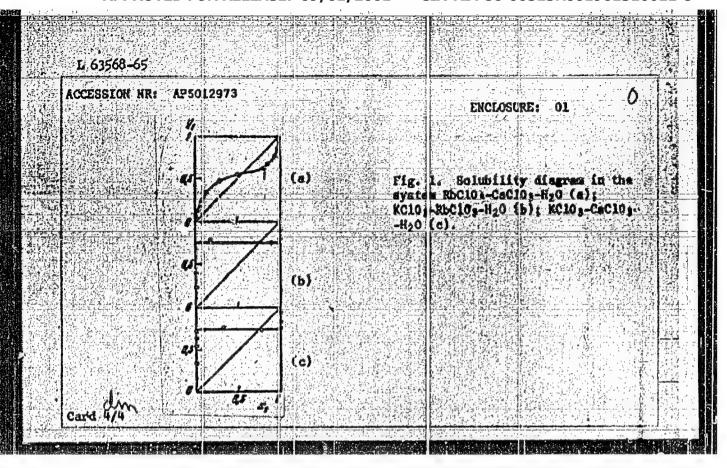
L 34503-65				
ACCESSION NR. AP500280				
ty and the fucion (amount)				0
ty and the fusion temperature compounds were stable to 800 posing somewhat more readily	if or LightO3 and F C. Both hydrolyz If than LightO3. C	TIO3 wer	e determine the K <sub>2</sub> TiO	i. Both 3 decom-
ASSOCIATION: None			us: 2 tables	and I figure.
SUBMITTED: 24Jul63	ENCL: 00	SUB	CODE: GC.	IC.
NR REF SOVI 005	OTHER: 016			
Card 2/2				
of Francisco and and an arrangement of the contract of the con		-		<b>新加州新州</b>

L 63568-65 EFA(*)-:/ENT(*)/EFF(*)/EFF(n)-2/T/ENF(t) Pt-7/Pu-1	
	#/0078/65/010/005/1225/1228 #1.123.62:546.135   I.: Korotkevich, B. 1.
TITLE: Ternary aquaous systems consisting of potassic chlorates at 25°C	11 11 11
SOURCE: Zhurnal neorganicheskoy khimii, v. 10, no. 5. TOPIC TAGS: potassium chlorate, rubidium chlorate, ce	
ABSTRACT: The solubility of KC103-CsC103-H2O, KC103-R -H2O ternary systems at 25°C is studied (see fig. 1 of tion of a fessible method for analysing potassium, ces discussed. The method of isothermal desupersaturation bility. In the first system, no solid solutions were rate and cesium chlorate, and only the liquid phase was system, both the solid and liquid phases were analyzed	the Enclosure). The selectum and rubidium systems is was used to study the solutormed between potassium chios analyzed. In the second
formed between potassism chlorate and rubidium chlorate colid and liquid phasen were also analyzed; rubidium of card 1/4	the True that the took are taken to the in the water

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SSSR (Institute of Inc	organic (hemi	stry, Siberian	Department,	Acudemy of	Sciences;
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"APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001961310011-8



### "APPROVED FOR RELEASE: 09/01/2001 CIA

CIA-RDP86-00513R001961310011-8

WILING, A.I.; BOCATHOTV, V.t.

Specific gravity of ion exchangers as dependent on their renie ecoposition. Znur. prikl. khim. 38 no.1:99-102 Ja 105.

(EPA 18:3)

VULIKH, A.I.; NIKOLAYEV, A.V.; ZAGOPSKAYA, M.K.; BOGATYREV, V.L. Absorption of ammonia and chlorine by ion-exchange resins under dynamic conditions. Dokl. AN SSSR 160 nc.5: 1072-107% P '65.

1. Institut neorganicheskoy khimii Sibirskogo otdeleniya AN SSSR. 2. Chlen-korrespondent AN SSSR (for Nikolayev).

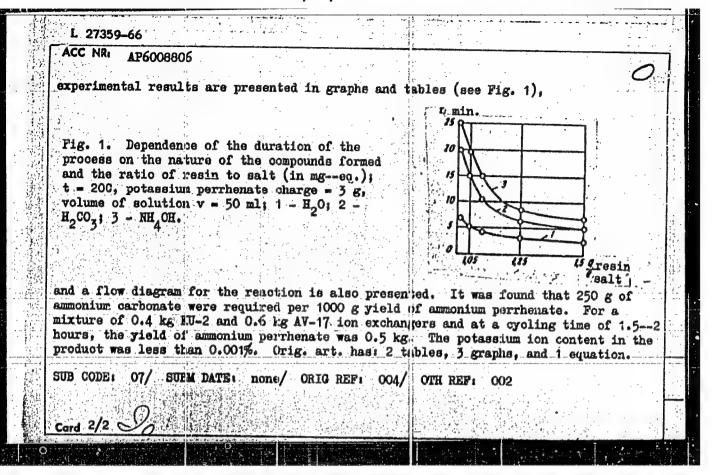
(MIRA 18:2)

KIRGINTSEV, A.N.; AVVAKUMOV, Yo.C.; WILIKH, A.I.

Behavior of alkali metal impurities in the crystallization of cesium salts from a melt. Dokl. AN SSER 164 no.6:1315-1318 0 165. (MIRA 18:10)

1. Institut neorganicheskoy khimli Sibirskogo otdeleniya AN SSSR. Submitted March 29, 1965.

ACC NRI AP6008806 SOURCE CODE: UR/0136/65/000/011/0096/0099 AUTHORS: Bogatyrev, V. L.; Vulikh, A. I.; Sokolova, S. I. ORG: none TITLE: Derivation of ammonium perrhenate from potassium perrhenate with the aid of mixed bed ion exchangers SOURCE: Tsvetnyye metally, no. 11, 1965, 96-99 TOPIC TAGS: ammonium salt, rhenium compound, ion exchange : esin, cation exchanger, anion exchanger, ion exchange/ KU-2 cation exchanger, AV-17 anion exchanger ABSTRACT: This investigation was conducted to extend the work of N. M. Rubinshteyn (Avt. svid. No. 148390 (Byull. izobret., No. 13, 1962)). Ammonium perrhenate was synthesized from potassium perrhenate and ammonium carbonate with the aid of a mixed bed KU-2 cation exchanger and AV-17 anion exchanger. The reaction was carried out according to the scheme  $RH + R'OH + KReO_4 = RK + R'ReO_4 + H_3O_1$  $2RH + R'_{2}CO_{2} + 2KReO_{4} = 2RK + 2R'R_{1}O_{4} + H_{2}CO_{2}(H_{2}O + CO_{2}t);$  $RNH_4 + R'OH + KReO_4 = RK + R'ReO_4 + NH_4OH (H_5O + NH_5†);$  $2RNH_4 + R'CO_8 + 2KReO_4 = 2RK + 2R'ReO_4 + (NH_4)_2 CO_8(H_2O + CO_8t + NH_3t).$ The optimum conditions for maximum yield of ammonium perrhenate were established. Card 1/2



EWT(1) L 34610-66 SOURCE CODE: UR/0210/66/000/003/0100/0102 ACC NR: AP6026571 AUTHOR: . Vulikh, A. I. (Candidate of technical sciences); Shivandronov, Yu. A. (Candidate of technical sciences); Zagorskaya, M. K. (Candidate of technical sciences); Bogatyrev, V. L. (Candidate of chemical sciences) ORG: Novosibirskiy Factory of Chemical Agents (Novosibirskiy zavod khimicheskikh reaktivov); Institute of Inorganic Chemistry, Siberian Branch, AN SSSR (Institut neorganiche skoy khimii Sibirskogo otdeleniya AN SSSR TITIE: Filtering ionite gas mask SOURCE: Gigiyena i sanitariya, no. 3, 1966, 100-102 TOPIC TAGS: gas mask, gas absorption, ion exchange resin, gas mask component, gas filter, industrial hygiene ABSTRACT: The authors tested in a wide range of concentrations and gas velocities the absorption from gas-air mixtures of armonia, amines (by KU-2 cationite in triesgen form), sulfur dioxide, chiorine, and hydrogen chloride (by AV-17 and EDE-10P anionites in the hydroxyl and carbonate forms). The basic and acidic gases were invariably completely absorbed. The capacity of the ionites was 80-90% of the total exchange capacity, 1.e., 4 meq/g for KU-2 and about 3 meq/g for AB-17. The most universal absorbents are the highly ionized single-function resins (KU-2, OBS-3, SEV, and AV. The carboxyl cationites (e.g., KB-4) and anionites with secondary and tertiary UDC: 614 **Card** 1/2

#### L 34610-66

ACC NR: AP6026571

amino groups (e.g., EDE-10P), whose capacity is 8-9 geq/kg, seem to be more effective in absorbing strongly acidic and strongly basic gases. Ionites with large pores (KU-2P for amines, etc.) are best for absorbing gases or fumes of organic substances with large molecules.

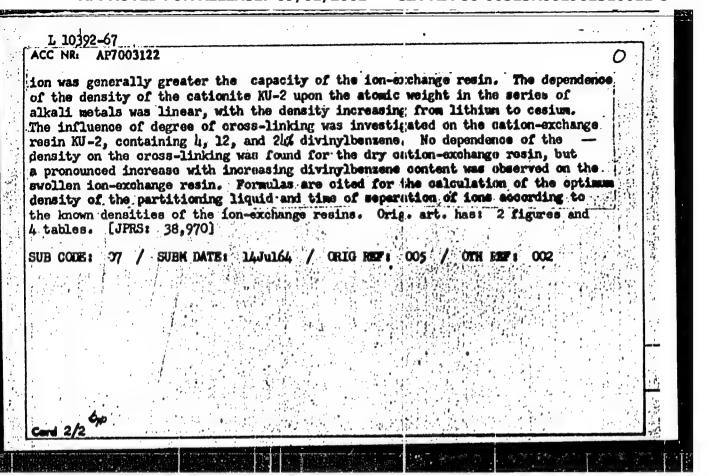
The article concludes with a brief description of an ionite gas mask successfully used for several menths under industrial conditions to provide protection against associate. An antidust filter from a RP-5 respirator is mounted on the lower part of the tank. KU-2 in the H form was the absorbent used. The total weight of the tank with the antidus: filter was 200-250 g. Loaded with 50 g of KU-2, it absorbed 3.5 g of associate and worked continuously for 30 hours. Orig. art. has: 1 figure and 1 table. [JPRS: 36,455]

SUB CODE: 06, 15, 07 / SUBM DATE: 24Nov64 / ORIG REF: 003 / OTH REF: 001

Card 2/2

L 05025-67 EWT(m)/EWP(t)/ETI IJP(c) JD/JG/WB ACC NRI AP6032980 SOURCE CODE: UR/0078/66/011/010/2328/2330 AUTHOR: Kirgintsev, A. N.; Avvakumov, Ye. G.; Vulikh, A. I. ORG: Institute of Inorganic Chemistry, Siberian Branch, AN SSSR (Institut neorganicheskoy khimii, Sibirskoye otdeleniye, AN SSSR) TITLE: Cesium nitrate purification by zonal recrystallization SOURCE: Zhurnal neorganicheskoy khimii, v. 11, no. 10, 1966, 2328-2330 TOPIC TAGS: metal crystallization, recrystallization, oriented crystallization, alkali metal, cesium nitrate, zonal recrystallization ABSTRACT: The method of oriented crystallization is used to determine the distribution of alkali metals in cesium nitrate at different crystallization rates (under constant mixing). The data obtained show that the method of zonal recrystallization may be recommended to free cesium nitrate of alkali metals. Orig. art. has: 1 table and 3 figures. [Authors' abstract] SUBM DATE: 08Jan65/ ORIG REF: 005/ SUB CODE: 07/ Card 1/1 UDC: 546, 36'175:548, 53

L 10392-67 EWT(m) DS/RH UR/0080/66, U39/008/1760/1765 SOURCE CODE ACC: NR: AP7003122 16 AUTHOR: Bogatyrev. V. L.; Vulikh, A. I.; Sokolova, S. I. ORG: Institute of Inorganic Chemistry, SO, AN SSSR (Institut neorganicheskoy khimii SO AN SSSR) TITIE: Density of ion-exchange resins SOURCE: Zhurnal prikladnoy khimii, v. 39, no. 8, 1366, 1760-1765 TOPIC TAGS: ion exchange resin, polymer cross linking ABSTRACT: A systematic determination was made of the densities of the most widespread industrial dation- and anion-exchange resins in various salt forms (in the dry and swollen states), for use in the development of technological and analytical methods based on the use of ion-exchange resins. The dependence of . the density of the investigated ion-exchange resing upon the nature of the sorbed ion, grain size, and degree of cross linking was dimenstrated. Fluctuation of the temperature within the range 10-30° was found to have no significant effect upon the results of the determinations. General patterns of variation were observed: 1) the density of the swollen ion-exchange resin was always less than the density of the dry resin, since the density of the latter was greater than one in all cases; 2) the density of the cation-exchange resins was generally greater in absolute magnitude than the density of the amion-exchange resins, which corresponds to the ratio of the densities of their matrices; 3) the density of the same ion-exchange resin increased with increasing equivalent weight of the sorbed ion; 4) the density of various ion-exchange resins containing the same UDC: 661.183.12 Card 1/2



ACC NR: AP6032947

SOURCE CODE: UR/0363/66/002/010/1803/1810

AUTHOR: Kutolin, S. A.; Vulikh, A. I.; Druz', N. A.; Shammasova, A. Ye.

ORG: none

TITLE: Dependence of the structure and properties of the A2BO3 and ABO3 compounds on the composition of the gaseous atmosphere in thermal

synthesis

SOURCE: AN SSSR. Izvestiya. Neorganicheskiya materialy, v. 2, no. 10, 1966,

1803-1810

TOPIC TAGS: ferroelectric material, antiferroelectric material, dielectric

constant, physical chemistry property, refractive index

ABSTRACT:

in a recently published article, the authors [association un-. known] analyzed the data from Western and Soviet literature, including their own experimental data which were published in 1964-66, on the thermal synthesis, structure, and properties of A2BO3 and ABO3 compounds, where A is an alkali metal and B is 'Ii, Zr, Mn, Nb, or Ta.

In previous publications, the authors established the effect of the gaseous medium in which the compounds were synthesized on their structure and particle size. Now, they have made a detailed analysis of the earlier data to correlate the conditions of synthesis,

Cord 1/5

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ACC NR: AP6032947

primarily the gaseous medium, with the physicochemical properties of the compounds. In the authors' opinion, this analysis is of practical importance for the synthesis and application of these compounds. The properties studied were: density, index of refraction, dielectric constant, intensity of IR absorption bands, and catalytic activity. The experimental data were obtained with samples sintered at a relatively low temperature from a solid mixture of an alkali carbonate and an acidic oxide, in vacuum or in a nitrogen stream.

The nature of the gaseous medium was shown to affect only the structure of alkali metatitanates and manganites (A<sub>2</sub>BO<sub>3</sub>), and not their physicochemical properties, such as density, index of refraction, or dielectric constant. Density was the only property of the manganites which was actually measured; the index of refraction and dielectric constant of the manganites exceeded the measurable values. An exception was the crystal symmetry of K<sub>2</sub>TiO<sub>3</sub> and RbTiO<sub>3</sub> which apparently remained unaffected by the gaseous medium in which their synthesis was accomplished. However, the existence in these two compounds of second order phase trunsitions, undetected by x-rays, may not be excluded. In all alkali metatitanates the intensity of the IR absorption bands due to deformation vibrations of

Card 2/5

CIA-RDP86-00513R001961310011-8

ACC NR. AP6032947

the [TiO<sub>6</sub>] octahedra was found to be independent of the method of synthesis. Their catalytic activity was affected by the gaseous medium, as shown, for example, by the comparative data on specific surface, preexponential factor, and activation energy for a maximum decomposition of hydrogen peroxide on a Li<sub>2</sub>TiO<sub>3</sub> catalyst prepared in the air or in vacuum.

In the group of A<sub>2</sub>BO<sub>3</sub> and ABO<sub>3</sub> compounds, where B is Zr, Nb, or Ta, i.e., alkali metazirconates, metaniobates, and metatantalates, only NaTaO<sub>3</sub> behaved like the alkali metatinates and manganites versus the gaseous atmosphere in the synthesis. The gaseous atmosphere changes the crystal structure, i.e., symmetry type and lattice constants of NaTaO<sub>3</sub>, but does not affect its picnometric density or intensity of deformation vibrational bands in their IR transmission spectra. Other compounds of this group -- Li<sub>2</sub>ZrO<sub>3</sub>, NaNbO<sub>3</sub>, KNbO<sub>3</sub>, CsNbO<sub>3</sub>, and CsTaO<sub>3</sub> -- change their crystal structure, i.e., symmetry type and/or lattice constant, in different gaseous media simultaneously with certain physicochemical properties, e.g., picnometric density, dielectric constant, intensity of deformation vibrational bands in the IR absorption spectra, and catalytic activity versus H<sub>2</sub>O<sub>2</sub> decomposition.

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ACC NR AP6032947

The crystal structure of LiNbO<sub>3</sub>, LiTaO<sub>3</sub>, and KTaO<sub>3</sub>, was not affected by the difference in gaseous atmosphere in the synthesis, but picnometric density, index of refraction, and intensity of deformation vibrational bands of the IR spectra were substantially changed.

These diverse and strong effects of the gaueous medium on the structure and properties of  $A_2BO_3$  and  $ABO_3$  compounds were explained as the result of deformability of their structure, specifically of the tendency toward distortion of the [TiO<sub>6</sub>], [NbO<sub>6</sub>], and [TaO<sub>6</sub>] octahedra. This deformability was correlated with a significant ionic polarizability of the alkali metatitanates, metaniobates, and metatantalates. This correlation which was experimentally established for the above-indicated compounds (presumably) may be extended to other compounds with significant ionic polarizability and may form the base for predicting the possibility of a beneficial effect of a given gaseous medium on the completeness of synthesis of a given compound. In addition, a significant ionic polarizability of a given compound may be an indication of a potential ferroelectric or antiferroelectric property.

An additional indication of the possible ferroelectric or antiferroelectric property of alkali metatitanates was seen in the ob-

ACC NRI A16032947

served analogy in the structure of their IR absorption bands which are linked to the stretching vibrations of the [TiO<sub>6</sub>] octahedra and in the structure of the corresponding bands of the [NbO<sub>6</sub>] and [TaO<sub>6</sub>] octahedra in the IR abosrption spectra of the alkali metaniobates and metatantalates. The observed spectral structure is characteristic of ferroelectric materials. The authors concluded that confirmation of the effect of a gaseous medium on solid-phase synthesis of a given compound is a prerequisite for studying the ferroelectric property in this compound. Orig. art. has: 1 figure and 8 tables.

SUB CODE: 11,07,20 / SUBM DATE: 14Jul65 / OI:IG REF: 022 / OTH REF: 016

Card 5/5

BOGATYREV, V.L.; VULIKH, A.I.; SOKOLOVA, S.I.

Obtaining ammonium perrhenate from potassium perrhenate with the help of a mixed layer of ionites. TSvet. met. 38 no.11: 96-99 N 165. (MIRA 18:11)

L 10846-66 EWT(m)/ETC/EWG(m) 4 RM/DS
ACC NR: AP6000233 SOURCE CODE: UR/0289/65/000/002/0023/0027
AUTHOR: Nikolayev, A. V.; Bogatyrev, V. L.; Vulikh, A. I., 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1,
(Institut neorganicheskoy khimii Sibirskogo otdeleniya AN SSSR)  TITLE: Separation of cation and anion exchanger in organic liquids
SOURCE: AN SSSR. Sibirskoye otdeleniye. Izvestiya. Seriya khimicheskikh nauk, no. 2, 1965, 23-27
TOPIC TAGS: anionite, ion exchange resin  ABSTRACT: The cation exchanger KU-2 was separated from the anion exchanger AV-17 in
mixtures of benzene, dichloroethane, and carbon tetrachloride. Values of the density and viscosity at 20C in these systems were determined. The dependence of the time of separation was studied as a function of the density of the separating liquid and grain size of the exchangers, and the effect of the difference in the density of the cation and anion exchanger during their separation was demonstrated. Formulas
derived earlier for the calculation of the optimum density of the separating liquid and duration of separation of cation and anion exchangers were confirmed experimentally. Orig. art. has: 3 figures and 5 tables.
SUB CODE: 07, 11 / SUBM DATE: 15Jun64  HW  Card 1/1  UDC: 541.13

KUTOLIN, S.A.; YULIKH, A.I.

Synthesis of metatitanates of alluline metals in a vacuum.

Zhur. ncorg. khim. 10 no.1:140-14 Ja \*65. (NIRA 18:11)

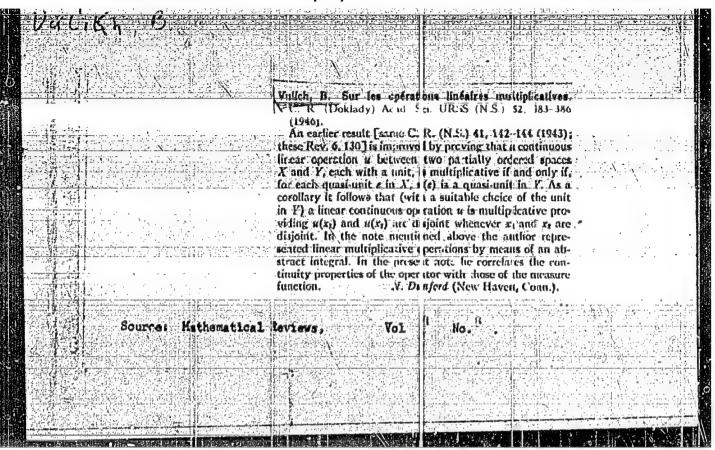
1. Submitted July 24, 1963.

VULTKH B.T.

Nekotoryve teoresy o posledovatel nostyakh razryvnykh funktsiy. DAN, 1 (1935), 357-362. Ob odnom tipe metricneskikh prostranstv. DAN, 4 (1915,) 295-298. Sur les forzes ge're'rals de certaines ope'rations line'aires. Matem. sb., 2 (44). (1937) 275-306. K hopmirovannyve prostranstva. L., Uche'n. zap. ped. in-ta, 28 (1939), 179-224.

So: Mathematics in the USSR, 1917-1947 adited by Kurosh, A.G., Narkushevich, A.I., Rashevshiy, P.K. Moscow-Leningrad, 1948

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Cources Mathem	ical Reviews   Yol -8		



Vulich, B. Sur quelquis operations num-infaires dans lease of capaca semi-ordonnés linéaires. C. R. (Doklady) Acad.  Sci. URSS (N.S.) 52  475-478 (1946).  This paper concerning the analytical representation of a certain type of nonlineix operator y with domain and range in partially ordered in tear vector's spaces X, Y satisfy, the axioms I-V of Kantorovitch).  The function u is partially additive if $u(x_1 + x_2) = u(x_1) + u(x_2)$ whenever $x_1$ and $x_2$ are disjoint; it is disjoint if $u(x_1)$ and $u(x_2)$ are disjoint; it is disjoint if $u(x_1)$ and $u(x_2)$ are disjoint venenever $x_1$ and $x_2$ are uniformly continuous if, for every constant C. limits $ u(x_1)-u(x_2) $ is zero in the (0-topology where $ x_1-x_2  < 6.1$ and $ x_2  =  x_2  < 6.1$ . A partially additive induction $ u(x_1)-u(x_2) $ is zero in the (0-topology where disjoint uniformly continuous operator is cilled an operator (ADC). If it addition the (0-convergence of $x_2$ to $x_3$ implies the (0)-convergence of $x_3$ to $x_4$ in $x_4$	VUINE DE			
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tion $u$ is uniformly continuous if, for every constant $C$ , function $(a \text{ measure function, respectively})$ and $Y(0) = 0$ , $ x_1 - x_2  < \delta \cdot 1$ and $ x_2  \cdot  x_3  < C \cdot 1$ . A partially additive tinuous on each finite segment, then for each $x$ in $x$ the disjoint uniformly continuous operator is called an operator $x$ in $x$ the $x$ in addition the $x$ -convergence of $x$ -convergence o	Cepaces semi-ordonn Sci. URSS (N.S.) 82 This paper concerns Certain type of nonline in partially ordered lid (the spaces X, Y satisf The function u is partial whenever x, and x;	(a linealres. C. R. (Doklady) Acad.  475-478 (1946).  the analytical representation of a constator u with domain and range sar vector spaces X, Y with a unit y the axioms I-V of Kantorovitch).  Ly additive if u(x <sub>1</sub> +x <sub>2</sub> ) = u(x <sub>1</sub> ) + u(x <sub>2</sub> ) are disjoint; it is disjoint if u(x <sub>1</sub> )	called unitary if $\varphi(e_1 + e_2)$ and if the $(o)$ -converge vergence of $\varphi(e_n)$ to $\varphi$ . In terms of these notice that general analytical $(ADC)$ respectively) $u(x) = \frac{1}{2} \frac{1}{2$	the stated without proof that it is given by the Radon interval.
Source: Mathematical Reviews, 1948, Vol 9, No. 1	tion $u$ is uniformly of $\lim_{t\to u}  u(x_1) - u(x_2) $ if $ x_1 - x_2  < \delta \cdot 1$ and $ x_1 $ disjoint uniformly continuous $(ADC)$ . If in addition t	intinuous if, for every constant $C_i$ of zero in the (o)-topology where I $ x_1  < C \cdot 1$ . A partially additive through operator is called an operator I is (o)-convergence of $x_i$ to $x$ implies	function (a measure full in a difficult $Y(\lambda) = \int_{0}^{\lambda} dx$	those values lie in $Y$ , $\varphi$ is a unitary suition, respectively) and $Y(0) = 0$ . $Z(t)dt$ , where $Z(t)$ is uniformly consequent, then for each $x$ in $X$ the
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VIIIIIKH KAREBUZU

Vulih, B. Z. The product in linear partially nintered spaces and its application to the fissory of operations. L. Mat. Sbornik N.S. 22(64), 27-18 (1948). (Russian) Vulih, B. Z. The product in linear partially ordered spaces and its application to the theory of operations. IL. Mat. Sbornik N.S 22(64), 267-317 (1948). (Russian) The author presents a detailed exposition, with a number of extensions, of results previously acnounced IC. R. (Doklarly) Acad. Sal. WRSS (N.S.) 16, 850-854, 855-859 (1940); 41, 142-144, 187-190 (1943), 52, 95-98, 383-380, 475-473 (1946); these Rev. 2, 221, 222; 6, 130, 8, 468; 9, 41]. Close connections exist between the theory expounded here and earlier results obtained by Freudenthal, Kantorovich, M. and S. Krein, and Kakutani [Freudenthal, Akad Wetensch, Amsterdaus, Proc. 39, 641-651 (1936); Kantorovitch, Rec. Math. [Mat. Sbornik] N.S. 2(44), 121-168 (1937); 7(49), 209-281 (1940); M. and S. Krein, Rec. Math. [Mat. Sbornik] N.S. 13(55), 1-38 (1943); Kakutani, Ann. of Math. (2) 42, 523-537, 994-1024 (1941); these Rev. 2, 317; 6, 276; 2, 318; 1, 205]. Let E be a partially ordered linear space over the real numbers (which are denoted by R in the sequel) of type  $S_0$ : i.e., (1) for some  $x \neq 0$ , OCE; (II) OCE and OCy imply OCE+2; (III) for every

rek, there exists an riel such that x1-x>0; (10) 0<x;

Source: Mathematical Reviews.

work, but 0 < a imply 0 < ax; (V) every subset of X bossolves above admits a least upper located. The author first proves a number of relations which obtain in all spaces of type Sci for example, if A is a bounted set in X and teX, then

inf  $(z, \sup_{x \in X} x) = \sup_{x \in A} \inf (x, z)$ .

No  $x_i$  a further restriction is placed upon the spaces considere i. A positive element in X, which may be denoted by the symbol  $1_i$  is said to be a unit if  $\inf(x, 1) > 0$  for all positive  $x \in X$ . It is supposed that X contains a unit element and t (at this unit is fixe t once and for all. An element  $x \in S$  is said to be unitary if all (t, 1-t)=0, and the set of all unitary elements in X is fenoted by  $\mathfrak{C}(X)$ . It is proved that unitary elements behave generally like projection operators in Hilbert space. For every  $x \in X$ , let  $e_i$  (the "characteristic" of x) to the least element in  $\mathfrak{C}(X)$  with the property that in  $\{\{i,i\}, 1-e\}=0$ ; it is shown that  $e_i$  exists for all  $x \in X$  and that  $e_i = \sup_{x \in X} \inf\{x_i \in X\}$ . It, Various other formal properties of  $e_i$  at also established.

Note; for all  $x \ge 0$ , let S(x) be the set of all states  $\sum_{i=1}^n a_i e_{i,k}$  where the as are assumptive real numbers,  $e_i e_i(X)$ , and the entire sum is less than of equal to x. For x, y in X and homographic, consider the set B of all sums  $\sum_{i=1}^n a_i e_i$  in  $(e_i, e_i)$ , where  $\sum_{i=1}^n a_i e_i e_i S(x)$  and  $\sum_{i=1}^n a_i e_i e_i S(y)$ . If B is bounded above the element suppose x is defined to be the product xy where  $a_i$  is defined to be the product  $a_i$  in the element  $a_i$  is  $a_i$  in the product  $a_i$  in the element  $a_i$  is an abounded, then the product  $a_i$  is  $a_i$  in the product  $a_i$  in the product  $a_i$  in the element  $a_i$  is  $a_i$  in the product  $a_i$  in the element  $a_i$  in the product  $a_i$  in the product  $a_i$  in the product  $a_i$  is  $a_i$  in the product  $a_i$  in the

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does not exist. For a and y nonpositive, he product to indefined as sup (x,0) sup (y,0)+(-1) inf  $(x,0)\cdot(-1)$  inf (y,0) — sup  $(x,0)\cdot(-1)$  inf (y,0)-(-1) inf (y,0) — (-1) inf (y,0) inf (y,0) inf (x,0) i four products exist; otherwise zy does rule exist. It is proved

inf (|=|, |y|)=0.

The airthin next considers the existence of inverse elements, as follows. If for in xext, there exists yell such that this is the case, then y is last to be inverse to be continuous functions. It is immediate that a section all the first of the first provided by x. See also il. Z. Valid, 2. 200. It is also proved that any space x can be indeed by x. It is immediate that a section all the first provided by x. It is immediate that a section all be indeed by x. It is immediate that a section all be indeed by x. It is immediate that a section all be indeed by x. It is immediate that a section all be indeed by x. It is immediate that a section all be indeed by x. It is immediate that a section all be indeed by x. It is immediate that a section all be indeed by x. It is immediate that a section all be indeed by x. It is immediate that a section all be indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section and x. It is immediate that a section all the indeed by x. It is immediate that a section all the indeed by x. It is immediate that a section x. It is a section x. rings are established, and it is shown has the product is a continuous function of both variable in the topology of o-convergence.

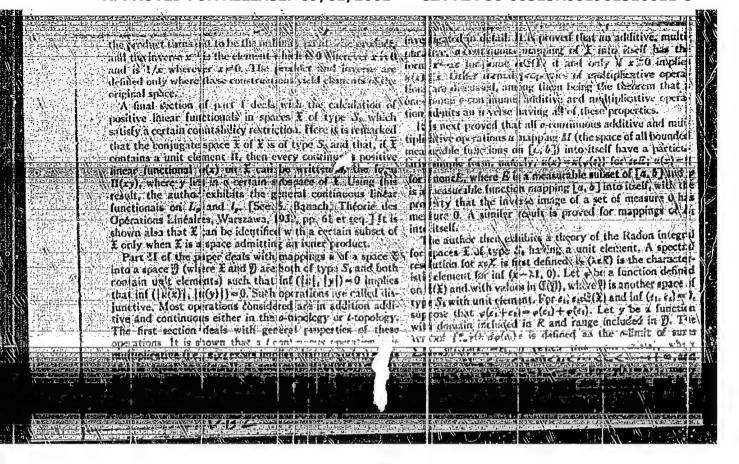
The next topic treated is that of spaces & which are ring under the operations it by and my. An element kee is said to be bounded if [2] if CI for some Call the set of all such a for a given nonnegative C is called a segment in K, and the set of all segments is denoted by K. Then Law clearly a space of type Si; is a ring; and it admiss the norm [2] sinf E[C, CeR, ] of SC. [3]. Then be results of since

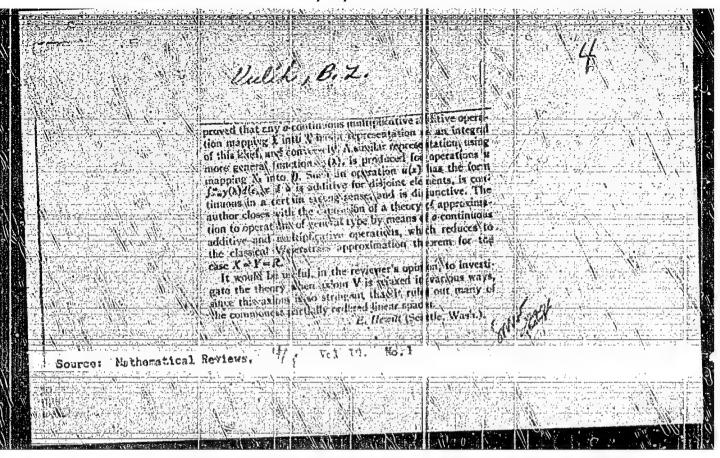
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the Ma is Mat. Shoralk N.S. 9(51); 3-26 (1941); these Rev. 1, 11, 2, is identifiable with a ring of real-valued continue & functions, defined on a certain compact Haus-dorll era in V. Elleviewer's pard: Xa is identifiable with the get of a feed valued continuous functions defined di Y if space Victor fact makes the introduction of multiplication in K to vial. [Reviewer's note: very similar considerations apply, if course, to Canich spaces satisfying Kakuthin's axioms for an Mapaco Dicacit, High of the interest of the prejent paper lies in the fact that multiplication is defined without recourse the representation theory.]

A in other of applications of the constructions jourlined above to discussed eg to L. I. and the space of all country by infinite sequender of real cumbers. In each case,

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YULIKH, B. Z. NOVOE 25337

Dokazatelstvo Odnoy Teoremy Kreynor. Uchen Zapiski (Danipr. Gos. Ped. In-T IM Gertsena) T. LXIV, 1948, S. 9-15

SO: LETOPIS NO. 30, 1948

natural despicion of multiplication done by a construction introduced by Kakutani [Ann. of Math. (2) 42, 523-537 (1941)) these Rev. 2, 3157 News it Asspace having a certain strong completeness property. is shown that every K-space can be imbedded in another It is also shown that every such complete K-space admits is first shown that any complete Boolean algebra tair serve ine Boolean algebra of components of a K-space. This is resiles, in para new, concerning extensions of & spaces It Booken sigebra is also proved. In chapter IV, we find that the tel of all components of a Kepace is a complete ermin special subspaces of K-spaces, colled components, are introduced. A linear subspace Z of a K-space X is called examponent if are, yeX, and Is Z y imply yeZ, and if of identities any crime 'v' 's ''. ''. ''. ''. and ''.'.''. bounded above in X. It is shown that every K-space X is standard examples of X-spaces are produced. In chapter II to us r-convergence, are defined and discussed and a fee and proved to ornvergence and "convergence here released Twenty different types of vactor littless (hipper I deals loss cit.] on integral representation of elements of X-spaces, sere considerably extended. A theorem of G. Birkhoff (Fraction Arnd. Ser. 11-2, A. 21, 151, 159, (10.3)), by the energy of the constant of the defining, change their axioms whenever convenient, and succeed in oint components. Chapter III presents Freudenthal gresults basic object the study of complete vector lattices. Such maticians in the field of functional analysis is covered in and two appendices. In the introduction there is an extended sicad. Sci. URSS (N.S.) 17 (1936 III), 9-14; Roc. Math. [Mat. Scornit.] M.S. 2(44), 171-165 (1937)], and many other writers. [For a basic list of original memoirs, cf. C. Birkhoff, Publ. v. curson of the theory of partially ordered linear spaces, as Birkhoff, Littice Teefry, Amer. Math. Soc. Collegnium Publ., v. 25, 2d ed., New York, 1948, those Rev. 10, 673.7 The Work is divided into an introduction thirrese chapters. a certain sense a subdirect sum of pair wise Proc. 39, 641-651 (1936) Kansartvic, 1. V., Vulin, B. Z., and Physics, A. G. Cosudarsty, ledat. Tehn. Teor. Lit. Moscow-Leningrad cional nyl analiz the volume under review contains an empelopedic dis in the course of the exposition, no fewer stouther according to the authors to the authors which the work of non-Soviet mathe 200 J. Kantorovič [e.g., C. R. (Deklady Lakad Wetensch Amsterdam for every pair of eleand the B . . . . Prostranstva lattice 718-

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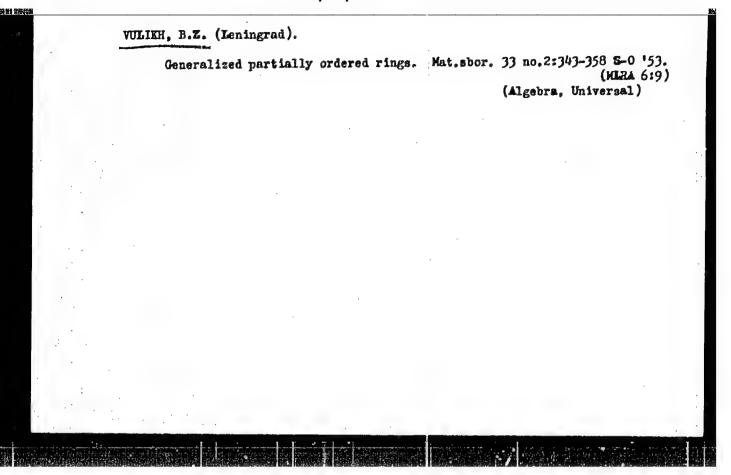
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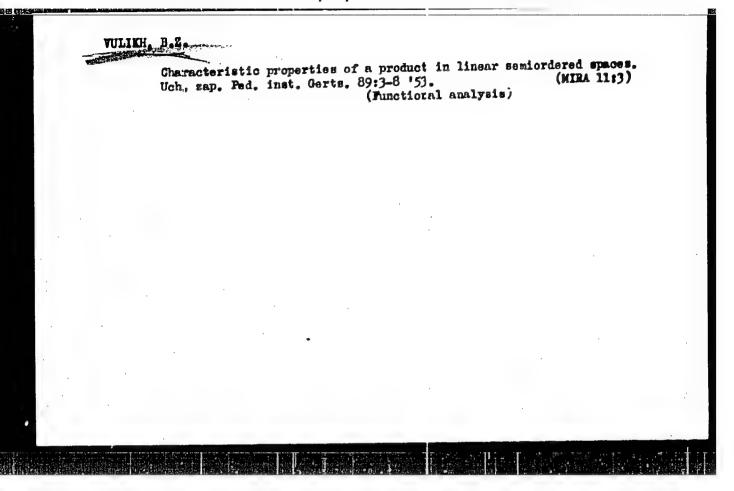
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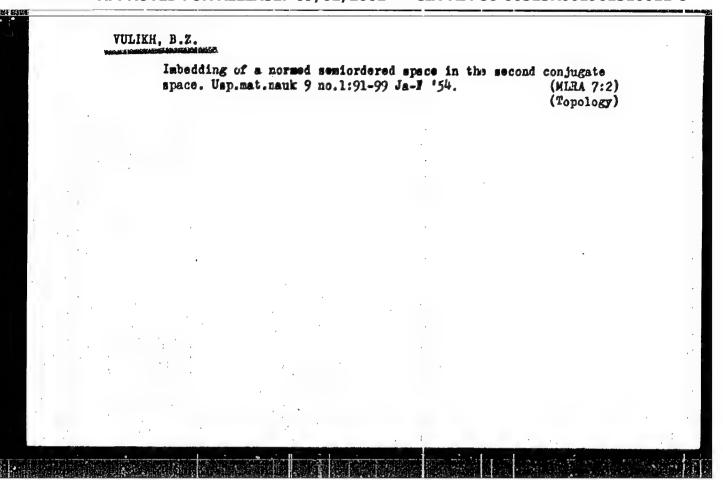
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SUBJECT

USSR/MATHEMATICS/Functional analysis

CARD 1/1 PG - 750

ROHTUA

VULICH B.Z.

TITLE

The application of the theory of partially ordered spaces to the investigation of selfadjoint operators in the Hilbert

PERIODICAL

Uspechi mat. Nauk 12, 1, 169-172 (1957)

reviewed 5/1957

The great analogy of the proofs and results of the theory of partially ordered spaces (lattice theory) and the theory of selfsdjoint operators induces the author to apply directly the theory of partially ordered spaces to the investigation of the selfadjoint operators. Without proof beside of some own results the author enumerates several well-known results of Stone, Sobolev, Ljubowin etc. Most of the results relate to bounded operators.

VULIMI, B.Z.

VULIKH, Boris Zakharovich

Partial order in rings of bounded self-conjugate operators [with summary in English]. Vest.IGU 12 no.13:13-21 '57. (MIRA 10:11) (Operators (Mathematics))

656

PHASE I BOOK EXPLOITATION

Vulikh, Boris Zakharovich

Vvedeniye v funktsional'nyy analiz (Introduction to Functional Analysis) Moscow, Gos. izd-vo fiziko-matematicheskoy lit-ry, 1956. 352 p. 7,500 copies printed.

Ed.: Akilov, G. P.; Tech. Ed.: Volchok, K. M.

PURPOSE: This book is intended for those interested in the fundamentals of functional analysis who do not have previous training in the more specialized branches of mathematics, and it may also be useful to engineers.

COVERAGE: The author gives only the fundamentals of functional analysis and their applications to problems in various fields of muthematics. Therefore the basic concepts of Euclidean, metric, normed, Hilbert and L spaces are given. A short theory of operators in various spaces, as well as their properties, are presented. As particular applications of functional analysis, the basic theorems of integral equations are presented. Various problems are studied, such as the problem of the best approximation, boundary value problem of differential equations, approximate methods of the solution of equations, generalized methods of the summability Card 1/10

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of series and others. There are 11 Soviet removed.	ferences; including 2 translations.
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